MULTIPLE CHOICE Choose the correct answer from the following choices:		
mixing technique		
separating technique		
boiling technique		
cooling technique		
2. Froth flotation process is used to concentrate the ore on:		
density basis		
concentration basis		
wetting basis		
magnetic basis		
3. Matte is a mixture of:		
FeS and CuS		
Cu ₂ O and FeO		
$Cu_2^{-}S$ and FeS		
CuS and FeO_2		
4. In the bessemerization process:		
roasted ore is heated		
molten matte is removed		
molten matte is heated		
molten matte is added		
5. Concentration of the copper ore is carried out by:		
calcinations		
roasting		
forth flotation		
distillation		

6. When CO, is passed through the ammonical brine the only salt that precipitates is:		
NaHCO ₃		
NH₄HCO ₃		
Na ₂ CO ₃		
$(NH_4)_2CO_3$		
7. In Solvay's process slaked lime is used to:		
prepare CO ₂		
prepare quick lime		
recover ammonia		
form Na ₂ CO ₃		
8. When NaHCO ₃ is heated it forms:		
CO ₂		
Ca(OH) ₂		
CaCO ₃		
CaO		
9. Formula of urea is:		
NH ₂ COONH ₄		
NH ₂ CONH ₄		
NH ₂ CONH ₂		
10. Crude oil is heated in the furnace upto:		
300°C		
350°C		
400°C		
450°C		

11. When heated crude oil is fed to the fractionating tower:		
vapours of higher boiling point fraction condense first in the lower		
part of the tower		
vapours of lower boiling point fraction condense first in the lower part		
of the tower		
vapours of higher boiling point condense later in the upper part of tower		
vapours of higher boiling point never condense		
12. Which one of the following is used as jet fuel:		
kerosene oil		
lubricating oil		
fuel oil		
diesel oil		
13. Which one of the following is not fraction of residual oil?		
paraffin wax		
asphalt		
fuel oil		
petroleum coke		
14. Which one of the followings is not a fraction of petroleum?		
kerosene oil		
diesel oil		
alcohol		
petrol		
15. The nitrogen present in urea is used by plants to synthesize:		
sugar		
proteins		
fats		
DNA		
16. Which one of the following organic compound is found in gasoline?		
C_2H_4		
C ₃ H ₈		
C ₈ H ₁₈		
$C_{12}H_{26}$		